

## A review on high performance liquid chromatography technique for pharmaceutical analysis

Khan Muhammed Azhar khan<sup>1</sup>, Prof.Mitali Dalwadi<sup>2</sup>, Dr.Umesh M.Upadhayay<sup>3</sup>

Author<sup>1</sup>, Co-Author<sup>2</sup>, Principle<sup>3</sup>, Department of Pharmaceutical Quality Assurance  
Sigma institute of Pharmacy, Bakrol, Vadodara-390019(Gujarat, India)

Date of Submission: 28-04-2024

Date of Acceptance: 08-05-2024

### ABSTRACT

HPLC is an analytical approach extensively used for identification, separation, spotting and quantification of various medicines and its affiliated degradants. Applicable mobile phase, stationary phase, column, column size, temperature, wavelength and grade selection is important for the suitable comity and stability of pharmaceutical as well as contaminants and degradants. Most of the medicines as well as other compounds can be assayed by HPLC method because of the several advantages like rapidity, specificity, accuracy, precision and ease of automation in this method. This article containing different aspects of Chromatographic technique including characteristics, instrumentation, important parameters and various applications of HPLC in different fields.

### I. INTRODUCTION

High-performance liquid chromatography or High pressure liquid chromatography (HPLC) is a specific form of column chromatography generally used in biochemistry and analysis to separate, identify, and quantify the active compounds. HPLC is the system of choice for checking peak purity of new chemical substances, covering reaction changes in synthetic procedures or gauge up, appraising new Pharmaceutical formulation and carrying out quality control/assurance of the final medicine products. HPLC is now one of the most important tools in analytical chemistry. It has the capability to separate, identify, and quantify the composites that are present in any sample that can be dissolved in a liquid. HPLC is the most accurate analytical approaches broadly used for the quantitative as well as qualitative analysis of medicine product and used for determining medicine product stability. In HPLC, the essential equipment consists of an eluent, reservoir, a high-pressure pump, and an injector for introducing the sample, a column containing the stationary phase, a detector and analyst. The development of broadly effective micro

particulate related phases has increased the versatility of the approach and has greatly enhanced the analysis of multi component syntheses.

### Principle<sup>[3-7]</sup>

HPLC principle is that solution of sample is injected into a column of passable material (stationary phase) and liquid phase (mobile phase) is pumped at high pressure through the column. The principle of separation followed is the adsorption of solute on stationary phase rested on its affinity towards stationary phase. HPLC is a special branch of column chromatography in which the mobile phase is forced through the column at high speed. As a result the analysis time is reduced by 1- 2 orders of magnitude relative to classical column chromatography and the use of much lower particles of the adsorbent or support becomes possible amplifying the column effectiveness mainly.

### Types of HPLC

#### 1. Normal Phase chromatography

In NP- HPLC the nature of stationary phase is polar and mobile phase is non-polar. In a combination of elements to be separated, those analytes which are fairly more polar will be retained by the polar stationary phase longer than those which are fairly less polar. Hence the least polar component will elute first. The attractive forces which are generally dipole-dipole and hydrogen bonding interaction. It is first choice for mixtures of isomers and for preparative scale HPLC and second choice for lipophilic samples that cannot dissolve well in water-organic mixtures.<sup>[8-9]</sup>

#### 2. Reversed Phase Chromatography

RP- HPLC has a non-polar stationary phase and polar or fairly polar mobile phase. RP-HPLC is grounded on the principle of hydrophobic interaction. In a combination of elements, those analytes which are moderately less polar will be retained by the non-polar stationary phase longer

than those which are moderately more polar. thus the most polar element will elute first. Molecules that retain some degree of hydrophobic character can be separated by reversed phase chromatography with excellent recovery and resolution.<sup>[10]</sup>

### 3. Size Exclusion Chromatography

SEC, also called as gel permeation chromatography or gel filtration chromatography mostly separates molecules on the base of size. The column is filled with material having precisely controlled notch sizes, and the sample is simply screened or filtered. Larger particles are quickly washed through the column; lower particles pierce inside the penetrable of the packing particles and elute afterward. This technique is comprehensively used for the molecular weight determination of polysaccharides.<sup>[11]</sup>

### 4. Ion Exchange Chromatography

In Ion- exchange chromatography, retention is grounded on the attraction between solute ions and charged spots bound to the stationary phase. This methodology is used nearly solely with ionic or ionizable samples. The stronger the charge on the sample, the stronger it'll be attracted to the ionic shell and therefore, the longer it'll take to elute.<sup>[11]</sup>

### 5. Bio-Affinity Chromatography

Separation Grounded on specific reversible dealings of proteins with ligands. Ligands are covalently affixed to solid support on abio-affinity matrix, retains proteins with dealings to the column- bound ligands.<sup>[12]</sup>

### Mode of separation<sup>[13]</sup>

There are two modes of separation in HPLC technique/method:

#### 1. Isocratic

In this mode the separation includes continuous eluent the composition and maintain the equilibrium condition in the column and the velocity of compound moving through the column at constant speed. Here the peak capacity is less and the longer component is stayed on the column the wider its peak.

#### 2. Gradient

In this type of mode, the separation contain different eluent composition, this help in increasing the separation power of the system mainly due to increase of the efficiency. Peak width

changes depending on the rate of the eluent composition variation.

### Instrumentation<sup>[14-15]</sup>

The required equipment consists of a high-pressure pump, and an injector for acquainting the sample, a column containing the stationary phase, a detector and recorder.

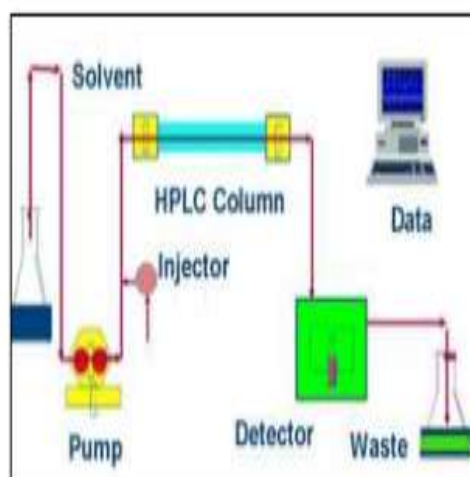


Figure:1 flow diagram of instrumentation of HPLC

### Injection of the Sample

Septum injectors are available; using which sample solution is injected. An injector (sample injector or autosampler) is suitable to introduce (inject) the sample into the continuously flowing mobile phase through that carries the sample into the HPLC column. A new advanced rotary cock and loop injector can be used to produce reproducible results. Typical sample volumes are 5- to 20- microliters ( $\mu$ l).

### Pump

A high- pressure pump (solvent delivery system or solvent reservoir) is used to induce and measure a specified inflow rate of mobile phase, generally milliliters per minute. The pump suctions the mobile phase from solvent reservoir and forces it to column and also passes to detector. The operating pressure depends on column confines, particle size, inflow rate and composition of mobile phase. Normal inflow rates in HPLC are in the 1 to 2ml/ min range. Typical pumps can reach pressures in the range of 6000- 9000 psi (400- to 600-bar).

### Columns

It's where the effective separation takes place. The column contains the chromatographic packing material required to effect the separation.

This packing material is called the stationary phase because it's held in place by the column tackle. It's stainless steel tube 5- 25 cm in length and 2-cm internal diameter. Material of packing is superficially passable or fully penetrable.

#### Detector

The detector can determine (discover) the individual particles that come out (elute) from the column. A detector serves to measure the volume of those particles so that the analyst can quantitatively analyze the sample elements. The detector provides an output to a recorder or computer that results in the liquid chromatogram (i.e., the graph of the detector response). There are several ways of detecting when a substance has passed through the column. Generally, UV spectroscopy is attached, which discovers the specific components. Multiple organic compounds absorb UV light of multi-coloured wavelengths. The volume of light absorbed will depend on the volume of a particular compound that's passing through the ray at the time.

#### System Suitability Parameters of HPLC

##### System Resolution

System resolution =  $2(t_2 - t_1) / (W_2 + W_1)$   
Where  $t_2$  and  $t_1$ , are the retention times of the two components and  $W_2$  and  $W_1$  are the corresponding peak widths. The resolution factor should be greater than 2.00 for two HPLC peaks.

##### Determination of System Precision

After a standard solution is injected a number of times, the relative standard deviation of the peak responses is measured as either the peak height or peak area. When using an internal standard method, the response ratio is calculated. Maximum allowable system related standard deviations made at the 99% confidence level have been tabulated. For the USP monographs, five replicate chromatograms are used if the stated limit for relative standard deviation is 2% or less. Six replicate chromatograms are used if the stated relative standard deviation is more than 2.0%.

##### Asymmetry Factor or Tailing Factor

The increase in the peak asymmetry is responsible for a falloff in chromatographic resolution, detection limits and precision. Measurement of peaks on solvent tails should be avoided. The peak asymmetry factor or tailing factor can

be calculated by following formula:  $T = W_{0.05} / 2f$ ,  
Where  $W_{0.05}$  is the width of the peak at 5 peak height.

##### Column Efficiency

Column efficiency is generally determined by calculating the number of theoretical plates for a column. It is mainly required for the assay of antibiotics and antibiotic containing drugs.

$$N = 5.545 (t / W_{h/2})^2 \text{ Or } N = 16(t / W)^2$$

Where  $t$  is the retention time of the analyte and  $W_{h/2}$  is the peak width at half-height or

$W$  is the width at the base of the peak. The height equivalent to one theoretical plate is calculated by  $h = L / n$  Where  $L$  is the length of the column.

##### Column Capacity (Capacity factor / Retention factor)

The column capacity factor is figured by  $K = (t_r - t_m) / t_m$ , Where the retention time of the solute is  $t_r$  and the retention time of solvent or unretained substance is  $t_m$ . Retention volumes are occasionally preferred, because they vary with flow rate. The factor is also calculated by  $V = V_r - V_m / V_m$ , Where  $V_r$  is the retention volume of the solute and  $V_m$  is the elution volume of an unretained substance.

#### Application of HPLC

##### HPLC industry Applications

There's a wide variety of applications throughout the process of creating a new medicine from drug discovery to the manufacture of pharmaceutical formulation that will be administered to patients.

##### Pharmaceutical applications:<sup>[18]</sup>

1. Tablet dissolution study of the pharmaceutical dosage form.
2. To control drug stability, Shelf-life determination.
3. Identification of active ingredients.
4. Pharmaceutical quality control.
5. Tablet dissolution of pharmaceutical dosage forms.

##### Food and Flavour analysis<sup>[17]</sup>

1. Rapid screening and analysis of components in nonalcoholic drinks.
2. Measurement of quality of soft drugs and water.
3. Sugar analysis in fruit juices.
4. Analysis of polycyclic compounds in vegetables.

5. preservative analysis.
6. Multiresidue analysis of lots of pesticides in food samples by LC triple quadrupole MS.

#### Forensics applications:<sup>[18]</sup>

1. Quantification of the drug biological samples.
2. Identification of anabolic steroids in serum, urine, sweat & hair.
3. Forensic analysis of textile dyes.
4. Determination of cocaine and other drugs of abuse in blood, urine, etc.
5. Determination of benzodiazepines in oral fluid using LC/MS/MS

#### Pharmaceutical drug discovery analysis

Developing a quickly, generic method for fast resolution Liquid chromatography with quadrupole MS detection. Fast, general LC/ MS method enables medicine analysis in lower than one minute.

#### Recent applications<sup>[19]</sup>

Analytic method development and validation are key elements of any pharmaceutical development program. HPLC analysis method is developed to identify, quantify or purifying compounds of interest. HPLC helps a lot in stability studies of drug formulations. HPLC helps a lot in stability studies of atropine, antibiotics, & biotechnology based drugs like insulin, streptokinase, etc.

1. It is used in inorganic chemistry for separating anions & cations.
2. It is used in forensic science for the separation of phenyl alkylamines (morphine and its metabolites) from blood plasma, and for the detection of poisons or intoxicants such as alcohol, carbon monoxide, cholinesterase inhibitors, heavy metals, hypnotics, etc.
3. It is used in environmental studies for analyzing the pesticide content in drinking water.
4. It is utilized in food analysis for separating water-soluble and fat-soluble vitamins from variety of food products, fortified food and animal feed.
5. It is also used for determining antioxidants and preservatives present in the food.
6. It is used in the cosmetic industry for the assay and quality control of various cosmetics like lipsticks, creams, ointments, etc.
7. It is used for separating various components of plant products with bear structural resemblance.
8. It is used in the agricultural industry for the separation of herbicides.

9. It is used in the separation and analysis of amino acids, carbohydrates, proteins, lipids and steroidal hormones.

10. It is used for separating coal and oil products from their crude sources.

11. It is used for separation and identification of Psychotropic drugs such as antidepressants, benzodiazepines, butyrophenones, neuroleptics, phenothiazines, etc.

12. It can be used for determining the stability of various pharmaceuticals. This is done by analyzing the degradation products of the drugs Eg: Stability studies of atropine

13. It can be used in bioassays of compounds like chloramphenicol, Cotrimoxazole, Penicillins, peptide hormones, and sulphonamides.

14. It is used for controlling microbiological processes used in the production of the number of antibiotics such as chloramphenicol, tetracyclines, and streptomycins.

15. It is used for monitoring the course of organic synthesis and also for isolating products in the reaction.

16. It gives an idea about the biopharmaceutical properties of a dosage form and the pharmacokinetics of the drugs. Thus, it is used in dosage form design.

17. It is utilized as an analytical method for numerous natural and synthetic drugs. It is used in different levels of pharmacy and pharmacology.

## II. CONCLUSION:

High-performance liquid chromatography is just the premier technique for Trace analysis of organic and inorganic composites. Determination of trace composites is veritably important in medicinal, biological, toxicology and environmental studies since indeed a trace substance can be dangerous or toxic. HPLC is applied for molecular weight determination, in logical chemistry, pharmaceutical and medicine science, clinical sciences, food technology, and consumer products, combinatorial chemistry, polymer chemistry, environmental chemistry and green chemistry. The role of HPLC in the pharmaceutical industry is veritably vital particularly in preformulation, process development, during formulation development and drug discovery and to corroborate medicine purity. All the work which has it being done in pharmaceutical substances, Preparation of pure composites, trace analysis, food safety where we've to analyze for fungicides and poisonous chemicals founds in food and food products all of these things are done routinely and

daily quickly by high performance liquid chromatography.

### REFERENCE

- [1]. Chawla G, Chaudhary KK. A review of HPLC technique covering its pharmaceutical, environmental, forensic, clinical and other applications. *Int J Pharm Chem Anal.* 2019;6(2):27-39.
- [2]. Martin, M.; Guiochon, G. Effects of high pressure in liquid chromatography. *J Chromatography A* 2005;1090(1):16-38.
- [3]. Dwivedia, S.K.; Agarwala, D.D. A Review: HPLC Method Development and Validation. *Int J Analytical Bioanalytical Chem* 2015;5(4):76-81.
- [4]. Rao, B.V.; Sowjanya, G.N.; Ajitha, A.; Rao, V.U. A review on stability indicating HPLC method development. *World Journal of Pharmacy and Pharmaceutical Sciences.* 2015;4(8):405-23.
- [5]. Taleuzzaman, M.; Ahmed, M.M.; Chattopadhyay, M. Particle size role, Importance and Strategy of HPLC Analysis-An update. *Int Arch BioMed Clin Res* 2016;1(2):3.
- [6]. Tamimi, L.; Dayyih, W.A.; Qinna, N.; Mallah, E.; Arafat, T. Pioglitazone HCl Levels and Its Pharmacokinetic Application in Presence of Sucralose in Animals Serum by HPLC Method. *Pharm Analytica Acta* 2014;5:318.
- [7]. Sultana, N.; Arayne, S.; Shah, S.N. Development and validation for the simultaneous quantification of prazosin, amlodipine, diltiazem and verapamil in API, dosage formulation and human serum by RP-HPLC: application to in vitro interaction studies. *Med Chem* 2014;4:770-7.
- [8]. Rao, G.; Goyal, A. An Overview on Analytical Method Development and Validation by Using HPLC. *Pharm Chem J* 2016;3(2):280-9.
- [9]. Synder, L.R.; Kirkland, J.J.; Glajch, J.L. In: *Practical HPLC Method Development*, 2nd ed; John Wiley and Sons Inc: Canada, 1997.
- [10]. Principles and Methods. In: *Amesham Biosciences of Reversed Phase Chromatography.* 1999;6-8.
- [11]. Types of HPLC. [http://hplc.chem.shu.edu/NEW/HPLC\\_Bo](http://hplc.chem.shu.edu/NEW/HPLC_Bo) ok/Introduction/int\_typs.html. (Accessed Feb 12, 2017)
- [12]. Malviya, R.; Bansal, V.; Pal, O.P.; Sharma, P.K. High performance liquid chromatography: a short review. *J Global Pharma Technol* 2010;2(5):22-6.
- [13]. Schellinger, A.P.; Carr, P.W. Isocratic and gradient elution chromatography: a comparison in terms of speed, retention reproducibility and quantitation. *J Chromatography A* 2006;1109(2):253-66.
- [14]. Malviya, R.; Bansal, V.; Pal, O.P.; Sharma, P.K. High performance liquid chromatography: a short review. *J Global Pharma Technol* 2010;2(5):22-6.
- [15]. McCown, S.M.; Southern, D.; Morrison, B.E. Solvent properties and their effects on gradient elution highperformance liquid chromatography: III. Experimental findings for water and acetonitrile. *J Chromatography A* 1986;352:493-509.
- [16]. Sneha, Lakshmi. R.P. A Review on Chromatography with High Performance Liquid Chromatography (HPLC) and its Functions. *Research and Reviews: J Pharml Analysis* 2015;4(1).
- [17]. Ma Bokai, Gou XinLei, Zhao XinYing, Application of highperformance liquid chromatography in food and drug safety analysis, *Journal of Food Safety and Quality*, 7, 2016, 4295-4298.
- [18]. Heewon Lee, Pharmaceutical applications of liquid chromatography coupled with mass spectrometry (LC/MS), *Journal of liquid chromatography & related technologies*, 28, 2005, 7-8.
- [19]. Sankar R, Snehalatha KS, Firdose ST, Babu PS. Applications in HPLC in pharmaceutical analysis. *International Journal of Pharmaceutical Sciences Review and Research.* 2019 Jan;59:117-24.